



Review

Shedding light on the mitochondrial permeability transition

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ABSTRACT

The mitochondrial permeability transition is an increase of permeability of the inner mitochondrial membrane to ions and solutes with an exclusion size of about 1500 Da. It is generally accepted that the permeability transition is due to opening of a high-conductance channel, the permeability transition pore. Although the molecular nature of the permeability transition pore remains undefined, a great deal is known about its regulation and role in pathophysiology. This review specifically covers the characterization of the permeability transition pore by chemical modification of specific residues through photoirradiation of mitochondria after treatment with porphyrins. The review also illustrates the basic principles of the photodynamic effect and the mechanisms of phototoxicity and discusses the unique properties of singlet oxygen generated by specific porphyrins in discrete mitochondrial domains. These experiments provided remarkable information on the role, interactions and topology of His and Cys residues in permeability transition pore modulation and defined an important role for the outer membrane 18 kDa translocator protein (formerly known as the peripheral benzodiazepine receptor) in regulation of the permeability transition.

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1. The mitochondrial permeability transition

The mitochondrial permeability transition (PT) is an increase of permeability of the inner mitochondrial membrane (IMM) to ions and solutes with an exclusion size of about 1500 Da [1–4]. The prevailing view is that the PT is due to opening of a regulated protein channel, the PT pore (PTP), which in the fully open state has an estimated diameter of about 3 nm [3,5]. PTP opening requires the presence of matrix $[Ca^{2+}]$,

Abbreviations: ANT, adenine nucleotide translocase; CP, coproporphyrin III; CRC, Ca^{2+} retention capacity; CsA, cyclosporin A; CyPD, cyclophilin D; Cu(OP)₂, copper-o-phenanthroline; DP, deuteroporphyrin IX; DTT, dithiothreitol; EGTA, [ethylenbis (oxoethylenetriolo)] tetraacetic acid; EP, etioporphyrin; FGIN1-27, N,N-dihexyl-2-(4-fluorophenyl) indole-3-acetamide; HP, hematoporphyrin IX; IMM, inner mitochondrial membrane; MBM⁺, trimethylammonium monobromobimane; MOPS, 4-morpholine-propanesulfonic acid; NEM, N-ethylmaleimide; ¹O₂, singlet oxygen; O₂^{•−}, superoxide anion; OMM, outer mitochondrial membrane; PhAsO, phenylarsine oxide; PK11195, 1-(2-chlorophenyl)-N-methyl-N-(1-methyl-propyl)-3-isoquinolinecarboxamide; PP, protoporphyrin IX; PT, permeability transition; PTP, permeability transition pore; RCR, respiratory control ratio; Ro5-4864, 7-chloro-5-(4-chlorophenyl)-1,3-dihydro-1-methyl-2H-1,4-benzodiazepin-2-one; ROS, reactive oxygen species; TEM, transmission electron microscopy; TSPO, 18 kDa translocator protein; UP, uroporphyrin I; VDAC, voltage-dependent anion channel

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which is an essential permissive factor, and of additional agents or conditions that are collectively termed “inducers” [6–15]. In spite of major efforts the molecular nature of the PTP remains undefined. Earlier candidates such as the IMM adenine nucleotide translocator (ANT) and the outer mitochondrial membrane (OMM) VDAC appear to be regulators rather than constituents of the PTP, as their genetic ablation did not prevent PTP opening [16–18] nor its sensitivity to cyclosporin A (CsA) [19–22], a high-affinity inhibitor of cyclophilins (CyP) that desensitizes the PTP to the inducing effects of Ca^{2+} and Pi through its binding to matrix CyPD [23–25]. Consistent with these findings, recent work has shown that the polyclonal antibody used to identify a CyPD-binding protein as ANT1 in fact labeled the Pi carrier, which has now been incorporated in the model of the PTP proposed by the Halestrap laboratory [12,26].

Ablation of the *Ppif* gene encoding CyPD has similar effects as CsA, in the sense that a higher of Ca^{2+} load and Pi is required for PTP opening in isolated mitochondria [27–30]. Pi is a classical inducer of the PTP, but we recently found that CyPD inhibition or its genetic ablation sensitizes the pore to the inhibitory effects of Pi, which would be the actual inhibitor of the pore under these conditions [31]. The evolutionary implications of this finding and their importance in reevaluating the yeast PTP (which is inhibited by Pi) as a model for the mammalian pore have been discussed in greater detail elsewhere [32].

1.1. The PTP as a fast Ca^{2+} release channel

Transient pore openings occur both in isolated mitochondria [33] and *in situ* [34], and we have proposed that the PTP could serve a

physiological function as a mitochondrial Ca^{2+} release channel [8,35]. This hypothesis is consistent with early results on the effects of CsA in Ca^{2+} distribution in rat ventricular cardiomyocytes [36] and with recent findings in *Ppif*^{-/-} mice whose heart mitochondria display a 2.6-fold elevation in total Ca^{2+} levels and consequent activation of intramitochondrial Ca^{2+} -dependent dehydrogenases resulting in stimulation of glucose oxidation at the expense of fatty acids [37]. Strikingly, CyPD ablation resulted in increased propensity to heart failure after transaortic constriction, overexpression of Ca^{2+} /calmodulin-dependent protein kinase II δ c and swimming exercise [37], findings that suggest a progressive Ca^{2+} overload due to decreased probability of PTP opening. That the PTP may serve as a physiological Ca^{2+} release channel is also supported by recent work in mouse primary adult neurons showing that PTP is activated in response to the combined action of more than one physiological stimulus affecting cytosolic $[\text{Ca}^{2+}]$ and that under these conditions PTP opening does not induce neuronal death but rather takes part in physiological Ca^{2+} dynamics [38].

1.2. Role of the PTP in disease

The role of the PTP as a mediator of cell death in paradigms relevant to human health is no longer questioned. A word of caution about an *exclusive* causative role of the pore is in order about early studies based on the effects of CsA alone because (i) inhibition of calcineurin prevents mitochondrial fission by inhibiting mitochondrial translocation of Drp1, a cytoprotective event that occurs independent of PTP inhibition [39]; and (ii) CyPD displays multiple regulatory interactions in mitochondria including CsA-sensitive binding to Hsp90 and TRAP1 [40], modulation of the F_0F_1 ATP synthase [41] and binding to Bcl2 [42] (see Ref. [43] for a review). Yet a role of the PTP is generally supported by studies of *Ppif*^{-/-} mice and of their cross-breeding with several disease genotypes.

Ppif^{-/-} mice were protected from ischemia-reperfusion injury of the heart [27,29] and brain [30,44], infarct size being reduced in CyPD-null adults compared to wild-type or littermate controls. These studies do corroborate the conclusions of previous studies based on the use of CsA in ischemic isolated hearts [45] and in infarcted patients [46], in brain damage by hypoglycemia [47], hyperglycemia [48], middle cerebral artery occlusion [49,50], and traumatic injury [51,52]. *Ppif*^{-/-} mice also display resistance to development of axonopathy in autoimmune encephalomyelitis [53] and to disease progression after crossing with superoxide dismutase 1 mutant mice [54], suggesting that CyPD-dependent (and possibly PTP-dependent) mechanisms are critical in the neurodegenerative aspects of demyelinating and motor neuron diseases. Ablation of CyPD substantially improved learning, memory, and synaptic function, and alleviated decreased long-term potentiation by the amyloid β peptide in a mouse model of Alzheimer's disease [55].

Perhaps the best case involving CyPD in pathology *in vivo* is represented by collagen VI diseases [56], a set of genetically heterogeneous conditions that cause Bethlem myopathy [57], Ullrich congenital muscular dystrophy [58,59] and myosclerosis [60]. The rodent model of the disease (*Col6a1*^{-/-} mice lacking collagen VI [61]) displays mitochondrial alterations in affected muscles that could be cured with CsA [62] or with the CyP inhibitor Debio 025 [63], and *Ppif*^{-/-} *Col6a1*^{-/-} mice lacking both CyPD and collagen VI became indistinguishable from syngenic wild type mice [64], a clear indication that CyPD is involved in the development of the disease downstream of the lack of collagen VI in a process that is amplified by defective autophagy [65]. These results match those obtained in patient cultures [66] and in a pilot clinical trial with CsA [67], and in *Ppif*^{-/-} *Scgd*^{-/-} and *Ppif*^{-/-} *Lama2*^{-/-} mice modeling sarcoglycan deficiency and congenital muscular dystrophy due to lack of laminin, respectively [68].

1.3. Modulation of the PTP by specific amino acid residues

As mentioned above, the molecular nature of the PTP is not known, yet we understand in some detail how the pore is modulated by key pathophysiological effectors like voltage, matrix pH and oxidative stress through specific Cys and His residues [8,10]. The PTP is voltage-dependent in the sense that the open conformation is favored by depolarization [69] while an acidic matrix pH locks the pore in the closed conformation through reversible protonation of His residues even in depolarized mitochondria [70,71]. The voltage-dependence in turn is modulated by redox events through a dithiol-disulfide interconversion at a matrix site that can be prevented by N-ethylmaleimide (NEM) [72] or monobromobimane [73]. A second redox-sensitive site whose oxidation increases the probability of PTP opening has been identified based on the effects of the IMM-impermeant reagent copper-*o*-phenanthroline ($\text{Cu}(\text{OP})_2$) [74]. Oxidation of this site promotes PTP opening that was inhibited by dithiotreitol (DTT) and β -mercaptoethanol but not monobromobimane [74]. Conclusive evidence that this site is located on the outer surface of the IMM rather than in the OMM was recently obtained in digitonin mitoplasts, which displayed an identical inducing effect of $\text{Cu}(\text{OP})_2$ as intact mitochondria [75].

Characterization of the PTP by chemical modification of specific residues has recently been pursued by a novel approach, i.e. photoirradiation of mitochondria after treatment with porphyrins [75–78]. The results of these experiments provided remarkable information on the role and interactions of His and Cys residues in PTP modulation, as well as on their topology, and suggested an important role for the 18 kDa translocator protein (TSPO), formerly known as the peripheral benzodiazepine receptor [79]. Before covering in greater detail the picture emerging from these studies, we will illustrate the basic principles of the photodynamic effect and the mechanisms of phototoxicity and discuss the unique properties of singlet oxygen ($^1\text{O}_2$) generated by specific porphyrins in discrete mitochondrial domains.

2. The photodynamic effect

Photosensitization is a process induced by highly noxious reactive species (free radicals and reactive oxygen species, ROS) which are generated by molecules (called photosensitizers) after they have adsorbed UV/visible light of appropriate wavelengths. When oxygen is required for photosensitization to occur, the overall process is defined as “photodynamic.” Photosensitization can lead to severe cell damage or death [80]. Nearly all organisms contain potential photosensitizers (e.g., bilirubin, chlorophylls and porphyrins), but under physiological conditions the photodynamic effect does not occur because the concentration of these compounds is low or because they are sequestered in complexes that inhibit photosensitization reactions. Photosensitization from these natural substances is only observed under extreme conditions such as the porphyrias, a group of diseases that causes severe skin lesions through overproduction of porphyrins. In addition to endogenous sensitizers, a broad variety of exogenous photosensitizers (present in food, cosmetics, plants or their juices, chemicals, drugs, etc.) can enter the body through ingestion, inhalation, injection or direct contact with the skin or exposed mucosa, causing photodamage in the presence of light.

Early last century scientists realized that the harmful effects of photosensitizers could be a tool for medicine [81,82]. Since then, the photochemical approach has primarily been developed as a treatment for cancer and for ophthalmologic and dermatologic disorders [83–88]. Although the term photodynamic therapy (PDT) was only defined at the beginning of the 20th century, this form of therapy can be dated back over 3000 years. In one of India's sacred books Atharva-Veda (1400 BC) a description is found of how

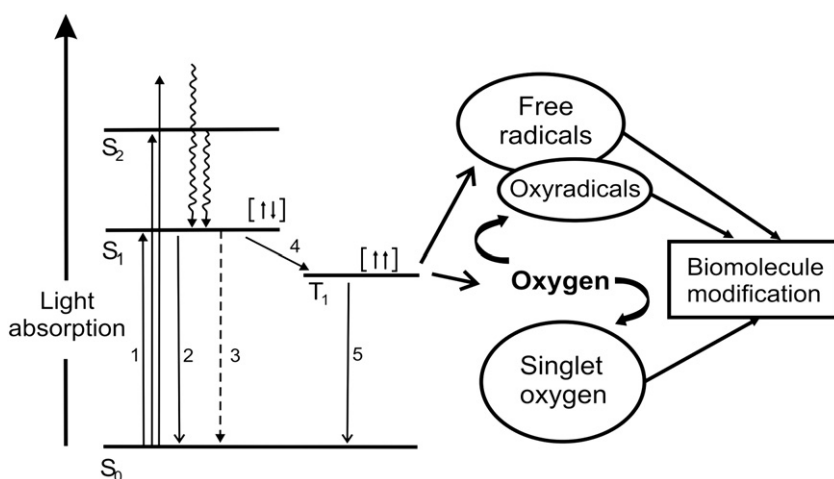


Fig. 1. Energy level diagram and photophysical and photobiological processes involving a photosensitizer. For explanation see text.

Psoralea corylifolia could be used for the treatment of vitiligo, which ancient Egyptians cured with extracts of *Ammi majus* growing on the banks of the Nile. The seeds of both plants contain psoralens, their photoactive components [81,82]. The modern history of PDT started with the observation that the combination of a chemical (acridine red) and visible light had a lethal effect on the microorganism *Infusoria*, a species of *Paramecium* [81,82]. The clinical potential of this discovery was not explored for several decades also because of the beginning of the antibiotic era. Today PDT (which inactivates many classes of pathogens) is a promising tool to overcome the problem of bacterial (multi-)drug resistance and to treat viral, fungal and parasitic infections [87,89,90].

2.1. Mechanisms of phototoxicity

The physical processes involved in sensitizer phototoxicity are illustrated in Fig. 1. The ground electronic state of the photosensitizer is a singlet state (S_0). On absorption of light of appropriate wavelength (1) the photosensitizer is excited to one of its short-lived singlet excited states, and rapid thermal relaxation leads to the lowest singlet excited state (S_1). The photosensitizer can then relax back to its ground state by fluorescence emission (2) and internal conversion (3), or can undergo intersystem crossing (4) to its lowest triplet excited state (T_1). This latter transition is spin-forbidden, but a good photosensitizer has a high probability of triplet-state formation after excitation. The T_1 -state is sufficiently long-lived to take part in chemical reactions, and therefore the photosensitizing action is mostly mediated by the T_1 -state. The T_1 -photosensitizer can also return to the S_0 -state by emitting phosphorescence (5). Activated photosensitizers in the excited triplet state can induce photochemical changes in a neighboring molecule via two competing pathways (Fig. 1). If oxygen is present in the environment, the photosensitizer can transfer its energy to form the highly reactive 1O_2 in a reaction defined as a type II mechanism [91]. The type I mechanism becomes important at low oxygen concentration or in a more polar environment. In this process the photosensitizer reacts directly with organic substrates, the solvent or another photosensitizer molecule via electron or hydrogen transfer to yield free radicals. These species are highly reactive and can easily interact with molecular oxygen to generate intermediates such as the superoxide anion ($O_2^{\cdot-}$), which can convert to H_2O_2 , the immediate precursor of the hydroxyl radical, the most dangerous member of the ROS family [91–93]. Both type I and type II reactions cause oxidation of cellular biomolecules. Frequently they occur simultaneously, and the ratio depends on

several parameters, the type of photosensitizer and the oxygen concentration being the most important [91,93]. For most photosensitizers employed in PDT, the type II photochemical reaction is the dominant process [94,95].

2.2. Singlet oxygen

In contrast to the vast majority of molecules, molecular oxygen in its ground state has an electronic triplet configuration with two unpaired electrons occupying two degenerate π antibonding orbitals (Fig. 2). In the two forms of excited singlet states ($^1\Sigma_g^+O_2$ and $^1\Delta_gO_2$) these electrons have opposite spin. The $^1\Delta_gO_2$ form (noted for simplicity as 1O_2) is involved in photosensitization, while $^1\Sigma_g^+O_2$ is too short-lived (10^{-11} – 10^{-9} s in solution) to be biologically relevant [93,96]. Electrophilic 1O_2 is seeking electrons to fill its vacant molecular orbital, and it thus reacts readily with electron-rich molecules. Targets for oxidation are lipids (cholesterol and phospholipids), amino acids (Trp, Tyr, His) and nucleic acid bases (guanine and guanosine) that have double bonds, as well as sulfur-containing amino acids (Cys, Met) [95,96]. In biological systems proteins are the main intracellular targets for 1O_2 owing to their abundance and fast rates of reaction, which are two orders of magnitude faster than those displayed with unsaturated lipids [97,98]. The reaction of 1O_2 with proteins can result in multiple effects including oxidation of side-chains, backbone fragmentation, dimerization/aggregation, unfolding

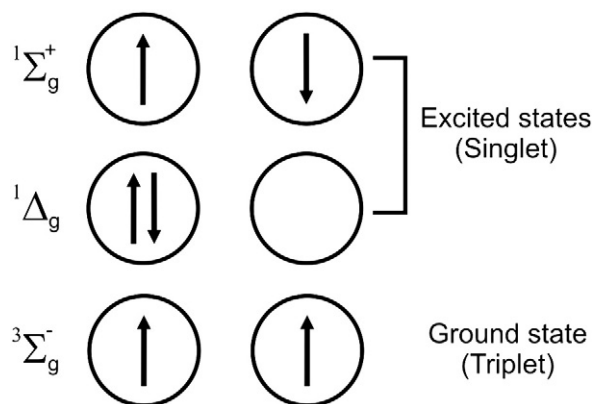


Fig. 2. Simplified representation of the electronic configurations of molecular oxygen lowest triplet and singlet states. The scheme depicts only π antibonding orbitals, for further explanation see text.

or other conformational changes, enzymatic inactivation, and alterations in cellular handling and turnover [97].

$^1\text{O}_2$ has unique features compared to other ROS. Being an electronically excited state it is generally produced when exogenous or endogenous photosensitizers absorb the appropriate wavelength of visible or UV-radiation; in contrast, other ROS such as H_2O_2 and O_2^- are produced in cells during normal physiological processes or are formed by activated neutrophils and macrophages. These “physiological” ROS can readily interconvert, making it difficult to discern which ROS is responsible for a particular cellular response, while $^1\text{O}_2$ does not interconvert to other ROS. Finally, the lifetime of $^1\text{O}_2$ has an inherent upper value (i.e. how long $^1\text{O}_2$ exists in the absence of reactive molecules) which poses an upper limit to its mobility from the site of generation. Based on the $^1\text{O}_2$ lifetime of $\sim 4\ \mu\text{s}$ a maximal diffusion radius of a mere $\sim 220\ \text{nm}$ in water has been calculated [98]. After this period of time $^1\text{O}_2$ returns to the ground state and loses reactivity by transferring its energy to the solvent. In contrast, other cellular ROS retain their activity until they hit a target or are destroyed by enzymes [98].

The lifetime of $^1\text{O}_2$ in homogeneous solutions is between 10^{-6} and $10^{-3}\ \text{s}$ [99]. This value is considerably shorter in cellular systems, ranging from 100 ns in the lipid regions of membranes to 250 ns in the cytoplasm. The shorter intracellular lifetime of $^1\text{O}_2$ indicates efficient quenching of this species by reaction with susceptible intracellular targets (for a review see Ref. [93]). Due to the abundance of these reactive molecules, the diffusion distance of $^1\text{O}_2$ in cells and tissues has been estimated to be in the order of $0.01\text{--}0.02\ \mu\text{m}$ [100–102]. Given a cell diameter of $10\text{--}30\ \mu\text{m}$, it can be reasonably assumed that the primary site of photodamage is also the site where $^1\text{O}_2$ has been generated. Thus, the intracellular localization of the photosensitizer is the major determinant in dictating which subcellular structures will be reached and attacked. Generally, mitochondrial photosensitizers are able to induce apoptosis very rapidly, lysosomal photosensitizers can elicit either a necrotic or an apoptotic response, and plasma membrane photosensitizers can lead to cell rescue or initiate apoptosis and/or necrosis (for a review, see Ref. [103]).

2.3. Porphyrins as photosensitizers

Porphyrins were identified in the mid-19th century, but it was not until the early 20th century that they were used in Medicine. Between 1908 and 1913 a number of photobiological experiments were carried out with hematoporphyrin (HP), demonstrating how it sensitized paramecia, erythrocytes, mice, and guinea pigs to light [81,82]. The pioneering report of human photosensitization by porphyrins was published in 1913 by Friedrich Meyer-Betz, who injected himself with 200 mg of HP and reported swelling and prolonged pain in light-exposed areas [82,104]. The subsequent discovery of the tumor-localizing properties of porphyrins and of the phototoxic effects on tumor tissues led to the development of modern clinical PDT [105–110].

The “photoactive core” of porphyrin-based compounds is the tetrapyrrolic macrocycle (porphine), while peripheral substituents control drug biodistribution and pharmacokinetics. Tetrapyrroles are naturally occurring pigments and vital constituents of oxidation–reduction and oxygen transport-related proteins such as hemoglobins, cytochromes, apoferritin, catalase, ferrichrome and peroxidases. Within these proteins tetrapyrroles are not able to induce photochemical reactions due to the presence of metal ions (Fe^{3+} , Al^{3+} , Mg^{2+}) which increase the probability of non-radiative decay of the triplet state. Indeed, the most efficient porphyrin-based photosensitizers (free base porphyrins) lack coordinated metal ions.

In the monomeric state free base porphyrins are potent photosensitizers with long lifetimes and high quantum yields of triplet

formation ($\Phi_T \sim 0.7\text{--}0.9$) while aggregation decreases the probability of photoexcitation. Because of the low energy gap between the triplet states of tetrapyrroles and molecular oxygen, a large fraction of triplets ($0.7\text{--}1.0$) leads to $^1\text{O}_2$ generation by energy transfer. Thus, $^1\text{O}_2$ is the predominant phototoxic species involved in porphyrin-photosensitized processes with a quantum yield of $^1\text{O}_2$ production (Φ_Δ) between about 0.7 and 0.9 for carboxylic porphyrins [111]. Alternative pathways for oxygen activation by photoexcited porphyrins (e.g. generation of O_2^- by electron transfer) usually have an efficiency lower than 0.1 [112].

Under the same experimental conditions, monomeric porphyrins exhibit quite similar photophysical properties and photosensitizing efficiency, independent of their chemical structure and the nature of the surrounding environment [111,113]. In particular, the yield of $^1\text{O}_2$ is similar for monomeric porphyrins with various degrees of hydrophobicity [111]. In spite of their similar basic structure and photophysical characteristics, however, porphyrins in cells display a wide range of activities even when they target the same organelle; and small differences (e.g. of peripheral substituents) result in different physicochemical properties that modulate the affinity for binding sites dictating uptake, subcellular distribution and therefore the final target of the photoprocess [111,114]. For these reasons a clever use of porphyrins can yield a wealth of information on biological processes.

2.4. Mitochondria as targets of porphyrin photodynamic activity

Studies with porphyrin-loaded cells indicate that hydrophobic porphyrins tend to concentrate in mitochondria, especially after prolonged incubation [115]. Mitochondrial damage is a major determinant of porphyrin phototoxicity [116–118] to which cells with few mitochondria are refractory [119]. Several mitochondrial components are affected by photoirradiation [115,116,119–121]. Light- and dose-dependence studies using dicarboxylic porphyrins (mainly HP) indicate that ATP synthesis is the first function to be lost because the F_0F_1 ATP synthase [115] and the adenine nucleotide translocator (ANT) [120] are the most vulnerable $^1\text{O}_2$ targets whereas respiration, Ca^{2+} uptake, OMM, matrix and intermembrane enzyme activities are more resistant in this order.

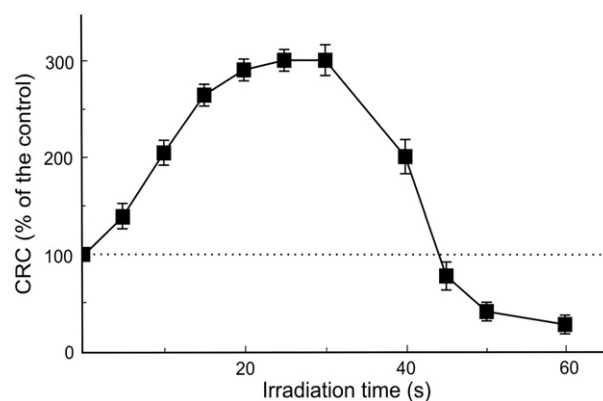


Fig. 3. Effect of increasing irradiation times on the Ca^{2+} retention capacity of HP-loaded mouse liver mitochondria. CD1 mouse liver mitochondria ($0.5\ \text{mg/ml}$) were incubated for 2 min at 25°C in a medium containing $250\ \text{mM}$ sucrose, $10\ \text{mM}$ Tris–Mops pH 7.4, $5\ \text{mM}$ succinate–Tris, $1\ \text{mM}$ P_i –Tris, $10\ \mu\text{M}$ EGTA–Tris, $3\ \mu\text{M}$ HP, $2\ \mu\text{M}$ rotenone, $0.5\ \mu\text{g/ml}$ oligomycin, $0.5\ \mu\text{M}$ Calcium Green-5N. After irradiation for the indicated periods of time at a fluence rate of $40\ \text{W/m}^2$, mitochondria were loaded with a series of $10\ \mu\text{M}$ Ca^{2+} pulses at 1 min intervals. PTP opening was determined as the Ca^{2+} retention capacity (CRC, expressed as the % of the CRC of non-irradiated organelles) measured at excitation $480\ \text{nm}$ and emission $530\ \text{nm}$. The CRC decrease at high doses of light was prevented by $1\ \mu\text{M}$ CSA (data not shown, see also Ref. [75]). Error bars refer to the mean \pm S.D. of four experiments.

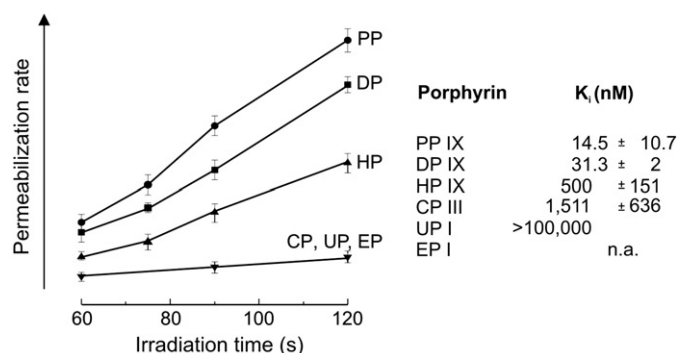


Fig. 4. Effect of PP-like (DP, HP, PP) and PP-unrelated (CP, EP, UP) porphyrins on light-dependent activation of the permeability transition. Wistar rat liver mitochondria (0.5 mg/ml) were incubated for 2 min at 25 °C in a medium containing 250 mM sucrose, 10 mM Tris–Mops, 5 mM succinate, 1 mM P_i –Tris, 10 μ M EGTA–Tris, 0.5 μ g/ml oligomycin, 2 μ M rotenone, pH 7.4 plus concentrations of each porphyrin giving a loading of 1.2 nmol of porphyrin/mg of protein (see also Ref. [75]). Preparations were then supplemented with 5 μ M Ca^{2+} (a concentration not sufficient to induce PTP opening per se) and irradiated for the indicated times at a fluence rate of 40 W/m². The PT was followed as the change in 90° light scattering at 540 nm. Error bars represent the mean \pm S.D. of three experiments. K_d values for porphyrins were taken from Ref. [123].

In 1992 Verma et al. [122] observed that the potencies of porphyrins at catalyzing cellular photodamage correlated closely with their affinities for the TSPO. Among a wide variety of porphyrins and porphyrin-like compounds the most photoactive were dicarboxylic porphyrins endowed with PP-like configuration such as PP itself,

mesoporphyrin IX, deuteroporphyrin (DP) and HP, which bind TSPO with nanomolar affinities [123]. Such selectivity for PP-like porphyrins has been ascribed to the major physiological role of TSPO to mediate mitochondrial uptake of naturally occurring dicarboxylic porphyrins such as PP, other heme precursors as well as heme itself [123,124]. TSPO is an 18 kDa OMM protein that may be located at junctional sites with the IMM [124–126]. Such a localization could influence substrate transfer between cytosolic and mitochondrial compartments, and porphyrins with the highest affinity for TSPO would be expected to accumulate best. The notion of TSPO mediating porphyrin phototoxicity was confirmed by our recent results in mitoplasts [75] while it has been challenged by other authors [127]. A key role of TSPO in controlling mitochondrial internalization and photoactivity of porphyrins is supported by the finding that high-affinity ligands of TSPO (such as 1-(2-chlorophenyl)-N-methyl-N-(1-methyl-propyl)-3-isoquinolinecarboxamide, PK11195, and 7-chloro-5-(4-chlorophenyl)-1,3-dihydro-1-methyl-2H-1,4-benzodiazepin-2-one, Ro5-4864) as well as cholesterol, prevent porphyrin-induced phototoxicity [128–131] probably by inhibiting translocation of porphyrins into mitochondria [129,131] (see also the following paragraphs).

3. Light on the PTP

In 1997 Salet et al. [76] and later Moreno et al. [77] exploited the photodynamic effect to study the PT. This approach provided detailed information on critical residues regulating the PT and helped to understand the role of IMM and OMM in PTP modulation [75–78]. The first observation was that a short photoradiation of mitochondria

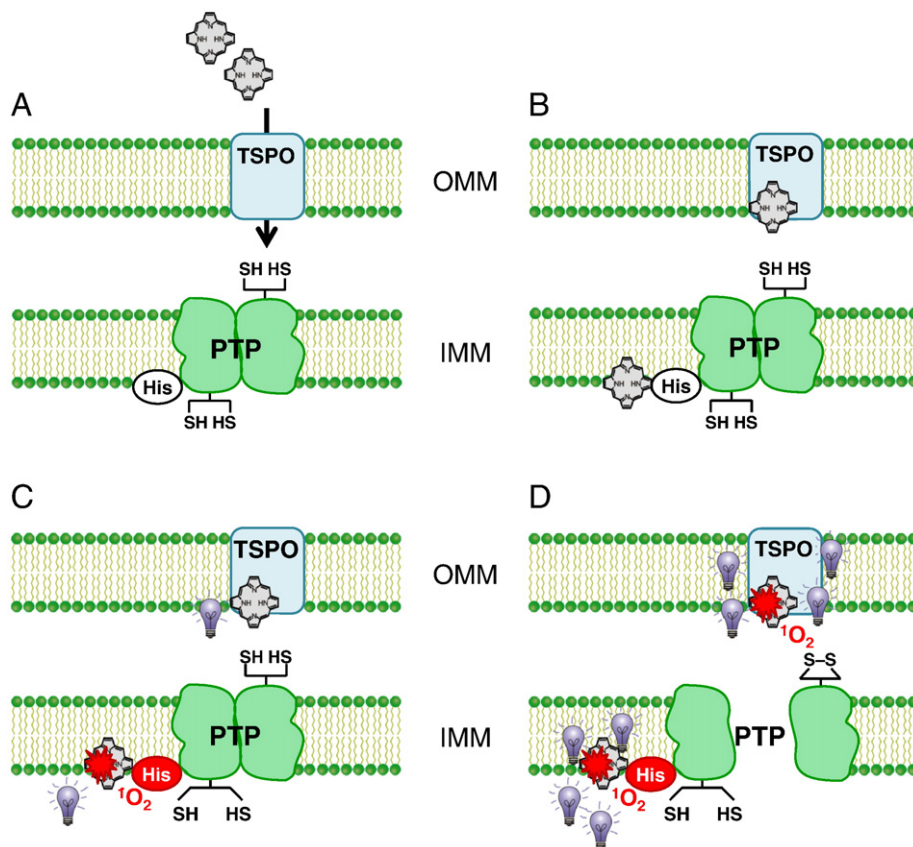


Fig. 5. Photodynamic events mediated by porphyrins at PTP-regulating His and Cys residues. (A) Porphyrins with PP-like configuration are transported inside mitochondria through TSPO. (B) Matrix-facing IMM porphyrin-binding sites are located in close proximity to critical His residues, and additional binding sites are adjacent to external thiols in close proximity to TSPO. In the dark, HP does not affect the structural properties of His- and Cys-containing domains, and PTP can open through selective oxidation of internal or external Cys when PhAsO or Cu(OP)₂ are added (omitted for clarity from the scheme, but see Refs. [75,78]). (C) After mitochondrial irradiation with moderate light doses, photoactivated HP generates ¹O₂ in the proximity of critical His residues which undergo modification and cause a structural rearrangement of the internal dithiol which can no longer undergo oxidation, thus resulting in PTP inhibition. (D) Irradiation with high light doses causes TSPO-dependent photoactivation of porphyrins in the OMM leading to oxidation of IMM surface thiols by ¹O₂, resulting in PTP opening. For further explanation see text.

loaded with HP caused inactivation of the ANT, yet mitochondria retained their ability to form a proton electrochemical gradient, and accumulated Ca^{2+} and Pi at the same rate as non-irradiated mitochondria. Strikingly, the oxidative effects of photodynamic action prevented opening of the PTP, which is normally induced by Ca^{2+} plus Pi, in the first example of pore inactivation by an oxidant [76], which is at variance with what is generally observed under oxidizing conditions [132–135]. The inactivating effect could be traced to photomodification of matrix His residues (the direct target of $^1\text{O}_2$) by vicinal porphyrin; this in turn lowered the reactivity of critical dithiols whose oxidation (e.g. by diamide) causes opening of the PTP [72]. Irradiation of HP-loaded mitochondria thus represents an example of site-selective inactivation of discrete pore functional domains comprising critical His and Cys residues in close structural and functional correlation [76].

The next finding was that irradiation with higher light doses caused instead opening of the PTP, which was attributed to photooxidation of IMM external thiols [75,78]. The effect of light dose on the Ca^{2+} retention capacity (CRC) of mouse liver mitochondria (a sensitive measure of the propensity of the PTP to open) is illustrated in Fig. 3. The initial increase of CRC (i.e. pore desensitization) was offset by a sharp decrease as the irradiation time increased. Since at high light dose the previously identified His and Cys residues have already been inactivated by matrix porphyrin, reactivation must depend on a different porphyrin site, which is specifically contributed by the OMM through TSPO [75]. The properties of this site are illustrated in the experiments of Fig. 4, where irradiation times of more than 60 s were applied to rat liver mitochondria preincubated with concentrations of PP, DP, HP, coproporphyrin III (CP), uroporphyrin I (UP) and etioporphyrin (EP) yielding an identical loading of 1.2 nmol of porphyrin $\times \text{mg}^{-1}$ of protein. The relative potency of PP-related porphyrins at sensitizing the PTP (measured here as the rate of mitochondrial permeabilization to sucrose) matched their affinity for TSPO, as based on the Ki displayed by each porphyrin for inhibition of binding of PK11195, a classical ligand of TSPO (see also Ref. [136]). The process of PTP activation was specific to porphyrins characterized by a PP-like configuration that bind TSPO with nanomolar affinity (PP, DP and HP), whereas PP-unrelated porphyrins such as tetracarboxylic CP, octacarboxylic UP and EP, which lacks carboxylic groups, were ineffective (Fig. 4 and Ref. [75]). It is also remarkable that porphyrin-mediated PTP photoactivation is suppressed (results not shown) by low concentrations of the high-affinity, porphyrin-competitive TSPO ligand FG1N1-27 [136].

The regulatory role of the OMM and TSPO on the PTP was confirmed by a study of mitoplasts prepared with digitonin concentrations that allowed selective removal of the OMM without interfering with mitochondrial energy-linked functions [75]. Like mitochondria, mitoplasts readily underwent PTP opening following Ca^{2+} uptake in a CSA-sensitive process, but major differences emerged in PTP regulation by ligands of TSPO. In mitoplasts the PTP could not be activated by photooxidation after treatment with dicarboxylic porphyrins endowed with PP configuration, and mitoplasts became resistant to the PTP-inducing effects of selective ligands of TSPO [75].

4. Summary and conclusions

In photosensitization of biological materials by $^1\text{O}_2$ the photo-damage is strictly limited to the immediate surroundings of the sensitizer because of the short diffusion distance and high reactivity of the photogenerated species. This peculiar oxidation mechanism provided novel information on residues regulating the PTP. Studies with this approach, whose results are summarized in Fig. 5, allowed to characterize the PTP-modulating properties of sites which exhibit different sensitivity towards oxidation by photoactivated porphyrins. Under basal conditions (Fig. 5A) the PTP favors the closed conforma-

tion. Porphyrins transported through TSPO are transferred to matrix, IMM and OMM sites, but in the absence of photoirradiation they do not affect the reactivity and pore-modulating properties of internal (matrix-exposed) and external cysteines located on the outer surface of the IMM [75] (Fig. 5B). Photoirradiation for short periods of time hits the most vulnerable site, which comprises matrix-exposed His and causes a secondary drop of reactivity of internal Cys, thus stabilizing the PTP in the closed conformation (Fig. 5C). Inactivation of internal Cys by His photodegradation allowed to establish the role of (i) external regulatory Cys, which can still undergo oxidation and thus increase the probability of PTP opening and (ii) of TSPO, which is essential for reactivation of the PTP by high light doses, an effect mediated in part at least by oxidation of regulatory thiols on the outer surface of the IMM (Fig. 5D).

TSPO thus fulfills a dual role in PTP modulation (i) as a *transport protein* for PTP-active compounds that are transferred to their PTP regulatory site(s) in the IMM or in the matrix and (ii) as a *PTP regulatory protein* when it binds its selective ligands like porphyrins. TSPO belongs to a family of ubiquitous proteins conserved from bacteria to mammals that bind small drugs, cholesterol, and porphyrins [79]. Consistent with its role in steroid hormone synthesis, in mammals TSPO is highly expressed in the adrenal cortex [137,138], and its inactivation induces an early embryonic-lethal phenotype in the mouse [139]. Establishing a role for TSPO in PTP modulation, which was also made possible by the unique features of the photodynamic effect, holds great promise for our continuing studies on the nature of the PTP, and on the development of drugs that can modulate its probability of opening.

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